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Code No: 121AE

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD B.Tech I Year Examinations, July - 2021

ENGINEERING CHEMISTRY

(Common to CE, EEE, ME, ECE, CSE, IT, MCT, ETM, AE, MIE, PTM)

Time: 3 hours Max. Marks: 75

Answer any five questions All questions carry equal marks

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1.a) b)	Describe the construction and working of Calomel electrode. What is Cathodic Protection? Explain sacrificial anode method.	[7+8]
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2.a) b)	What is an electroless plating? Explain the electroless plating of nickel. Define fuel cell? Explain the construction and working of Methanol-Oxygen fue	el cell
0)	Define raci cent. Explain the construction and working of including Oxygen rac	[7+8]
3.a)	Write the mechanism of chain growth polymerization with suitable examples.	
b)	Explain the preparation of nanomaterials by sol-gel method.	[7+8]
4.a)	Discuss various mechanisms of lubrication with examples.	
b)	Discuss the various steps involved in setting and hardening of cement.	[7+8]
5.a)	Give the differences between scales and sludges.	
b)	Describe the Ion-exchange process for softening of hard water.	[7+8]
6.a)	What is Caustic-embrittlement? Explain its disadvantages and prevention.	
b)	Describe the Reverse Osmosis process for softening of hard water and	give its
	advantages.	[7+8]
7.a)	Explain Proximate analysis of coal. What is its significance?	
b)	What is Petrol? How is it synthesized by Fischer-Tropsch's process?	[7+8]
8.a)	Describe the phase diagram of one-component water system.	
b)	Derive Langmuir adsorption isotherm. Give the applications of colloids in industry.	
		[7+8]

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